



BaNiSn₃-type ternary germanides SrMGe₃ (M = Ir; Pd and Pt)

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ABSTRACT

Ternary intermetallics SrMGe₃ (M = Rh, Ir, Ni, Pd and Pt) have been prepared by arc melting. The crystal structure of these intermetallics was determined by single-crystal X-ray diffraction for M = Ir, Pd and Pt. The structure of those three intermetallics is BaAl₄-derivative BaNiSn₃-type with the space group *I4mm* (No. 107). The other two, Rh and Ni, are likely to take the same structure, speculated from the corresponding XRD patterns using the powder samples. Neither of them shows superconductivity down to 1.8 K.

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1. Introduction

Among ternary intermetallic compounds, ThCr₂Si₂-type intermetallics have been extensively studied, especially for the interest of the superconducting and magnetic properties. The structure of ThCr₂Si₂ is the ordered ternary derivative of the binary BaAl₄-type structure [1]. Although superconductivity is observed for some compounds, the critical temperature (*T_c*) is generally very low, as reported for LaPd₂Ge₂ and LaIr₂Ge₂ with *T_c*'s of 1.12 [2] and 1.5 K [3], respectively. Many works were carried out for the discovery of new intermetallic superconductors with higher *T_c*'s. Finally, quaternary intermetallic superconductors AT₂B₂C with ThCr₂Si₂-derivative structure showing high *T_c*'s were discovered [4–7]. Among these borocarbides, YPd₂B₂C shows the highest *T_c* of 23 K. Furthermore, recently new ThCr₂Si₂-type superconductors (Ba,Sr)_{1–x}(K,Cs)_xFe₂As₂ were discovered with *T_c*'s up to 38 K [8,9]. Thus, this structure is one of the keys to search for new high-*T_c* intermetallic superconductors. Indeed, we have recently discovered ThCr₂Si₂-type ternary germanide SrPd₂Ge₂ shows superconductivity with a *T_c* of 3.0 K [10]. Other related germanides SrM₂Ge₂ (M = Ni and Ir) with the same ThCr₂Si₂-type structure do not show superconductivity down to 1.8 K [11].

Numerous intermetallics with this ThCr₂Si₂-type structure have been reported. The ThCr₂Si₂-type structure is one of the derivatives of the binary BaAl₄-type structure, as mentioned above. The inter-

metallics with other derivative structures such as CaBe₂Ge₂ and BaNiSn₃ have been much less reported. The structural difference among these three types in A–M–X ternary systems is as follows: XMX–XMX in ThCr₂Si₂, MXM–XMX in CaBe₂Ge₂, MXX–MXX in BaNiSn₃. In this paper we report new BaNiSn₃-type intermetallics SrMGe₃ (M = Ir, Pd and Pt). Neither of them shows superconductivity down to 1.8 K.

2. Experimental

Starting materials were Sr (sheet, 99% in purity), Rh (powder, 99.9%), Ir (powder, 99.9%), Ni (shot, 99.9%), Pd (sheet, 99.95%), Pt (sheet, 99.95%) and Ge (grain, 99.99%). The Rh and Ir powders were preliminary arc melted to form small grains. They were arc melted with a stoichiometric ratio of SrMGe₃ (M = Rh, Ir, Ni, Pd and Pt) under Ar gas atmosphere on a water-cooler copper hearth. The melting was repeated several times with the button turned over between each melt. First M and Ge were arc melted, and then the melted buttons were melted together with Sr in order to minimize the loss of Sr. The weight loss after the melting was within a few percent. The obtained buttons wrapped in a Ti foil were annealed in an evacuated silica tube at 1173 K for one week and quenched into a cold-water bath.

Phase identification was carried out for crushed samples by an X-ray diffraction (XRD) method with an X-ray diffractometer Rigaku RINT-TTR III with monochromatized Cu K α radiation. Single crystals for structural analyses picked up from the crushed samples were glued on the top of a glass fiber and mounted on the goniometer head. X-ray single crystal diffraction data were collected at room temperature 293(2) K using a Bruker SMART APEX CCD area-detector diffractometer with graphite monochromatized Mo K α radiation (λ = 0.071073 nm). The absorption correction and structural refinement were carried out using the programs SADABS and SHELXL97 [12–14].

Microstructural observation was carried out using a scanning electron microscope (SEM) JEOL JSM-6301F with an energy dispersive X-ray (EDX) spectrometer.

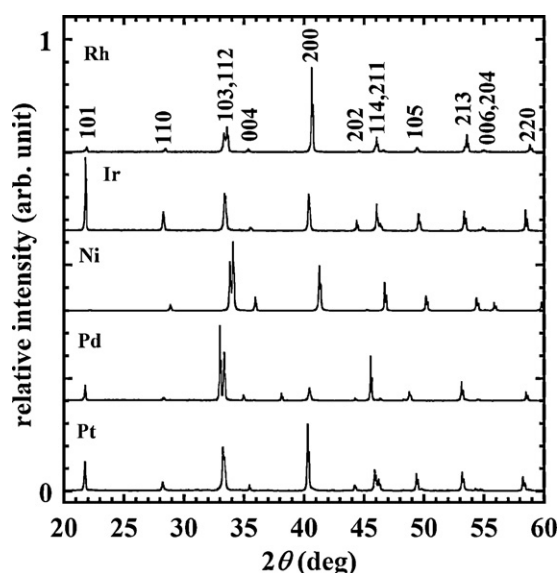
DC magnetization measurements were performed for polycrystalline bulk samples at temperatures above 1.8 K with a superconducting quantum interference device (SQUID) magnetometer Quantum Design MPMS XL. Each measurement was carried out for a few samples to check reproducibility.

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Table 1Atomic coordinates and equivalent thermal parameters for BaNiSn₃-type SrMGe₃ (M = Ir, Pd and Pt).

Site	Atom	Wyckoff position	x	y	z	$U_{eq} \times 10^4$ (nm ²)
Sr	Sr	2a	0	0	0.0000(3) ¹	1.49(4) ¹
					0.0000(2) ²	1.45(3) ²
					0.0000(2) ³	1.50(3) ³
M	M	2a	0	0	0.6493(1) ¹	1.21(2) ¹
					0.6419(1) ²	1.23(2) ²
					0.6448(1) ³	1.30(1) ³
Ge1	Ge	4b	0	1/2	0.2537(2) ¹	1.52(4) ¹
					0.2532(1) ²	1.34(2) ²
					0.2570(1) ³	1.41(3) ³
Ge2	Ge	2a	0	0	0.4123(2) ¹	1.55(5) ¹
					0.3994(2) ²	1.43(4) ²
					0.4012(2) ³	1.38(4) ³

Superscripts 1, 2 and 3 indicate the parameters for M = Ir, Pd and Pt, respectively.

**Fig. 1.** XRD patterns of crushed polycrystalline samples of SrMGe₃ (M = Rh, Ir, Ni, Pd and Pt). Most of the XRD peaks are indexed on the basis of a tetragonal unit cell.

3. Results and discussion

Fig. 1 shows the XRD patterns of crushed powder sample SrMGe₃ (M = Rh, Ir, Ni, Pd and Pt). Most of the diffraction peaks in the XRD patterns are indexed on the basis of a tetragonal unit cell. The reflection condition of $h + k + l = 2n$. This indicates that the lattice is body centered with one of suggested space groups $I\bar{4}m2$, $I\bar{4}2m$, $I4mm$, $I422$ and $I4/mmm$. The lattice parameters of SrIrGe₃ are $a = 0.446$

and $c = 1.010$ nm. The SrMGe₃ with the other M's show the lattice parameters almost equal to those of SrIrGe₃. The color of SrRhGe₃ and SrIrGe₃ is slightly red, the same as ThCr₂Si₂-type SrNi₂Ge₂ and SrPd₂Ge₂, whereas the other intermetallics SrNiGe₃, SrPdGe₃ and SrPtGe₃ are metallic gray. Preliminary experiment revealed that arc melting of SrMGe₃ (M = Fe, Ru, Os and Co) did not form this tetragonal phase at all.

SEM–EDX analyses were performed on the polished cross-section of the buttons of SrMGe₃. The metallographic observation indicated that these samples were nearly single-phase, in agreement with the XRD patterns shown in Fig. 1. The composition of the main phase in these samples is Sr:M:Ge = 1:1:3, indicating that the phase with the tetragonal cell is indeed SrMGe₃.

Shiny crystals picked up from crushed samples of M = Ir, Pd and Pt were used for structural refinement. Preliminary investigations indicated that these crystals had almost the same lattice parameters as the polycrystalline SrMGe₃. The lattice parameters of SrMGe₃ are almost equal to those of LaIrGe₃ with $a = 0.4428(4)$ and $c = 1.0042(9)$ nm and BaPtGe₃ with $a = 0.45636(2)$ and $c = 1.023416(6)$ nm [15,16]. The structure of these intermetallics is BaNiSn₃-type with the space group $I4mm$ (No. 107). Therefore, the structural refinements were firstly carried out on the basis of the model of BaNiSn₃-type structure for SrMGe₃.

Table 1 lists the atomic coordinates and equivalent thermal parameters for SrMGe₃ (M = Ir, Pd and Pt). The crystal data and the results of structural refinements are listed in Table 2. The small R factors for these three intermetallics indicate that the structure of SrMGe₃ is BaNiSn₃-type. The difference of the number of refined parameters in Table 2 is associated with the extinction effect. The effect was ignorable for both SrPdGe₃ and SrPtGe₃ samples. Fig. 2 shows the crystal structure of BaAl₄, together with its derivative BaNiSn₃-type SrMGe₃ and ThCr₂Si₂. For the other two, SrRhGe₃

Table 2Crystal data and results of the structural refinements for BaNiSn₃-type SrMGe₃ (M = Ir, Pd and Pt).

Compound	SrIrGe ₃	SrPdGe ₃	SrPtGe ₃
Space group		$I4mm$ (No. 107)	
Formula units/cell		$Z = 2$	
Calculated density (gcm ⁻³)	8.213	6.685	8.178
Crystal size (μm ³)	40 × 40 × 100	60 × 100 × 100	60 × 80 × 160
Absorption coefficient (mm ⁻¹)	68.031	38.81	69.020
R_{int}	0.0288	0.0315	0.0518
$2\theta_{max}$	80.48	80.42	80.84
Number of measured reflections	2262	2271	2140
Number of unique reflections	374	415	417
Number of reflections with $I > 2\sigma(I)$	374	413	415
hkl range	$-8 \leq h \leq 8, -7 \leq k \leq 8, -12 \leq l \leq 18$	$-8 \leq h \leq 7, -8 \leq k \leq 8, -18 \leq l \leq 17$	$-7 \leq h \leq 8, -7 \leq k \leq 7, -18 \leq l \leq 18$
Number of refined parameters	15	14	
Final residual $R[F^2 > 2\sigma(F^2)]$	0.0401	0.0431	0.0342
Weighted $R(F^2)$	0.1106	0.1018	0.0825
Goodness of fit on F^2	1.101	1.119	1.104
$\Delta\rho_{max}/\Delta\rho_{min}$ (e/Å ³)	3.001/−4.508	2.434/−1.585	4.380/−2.319

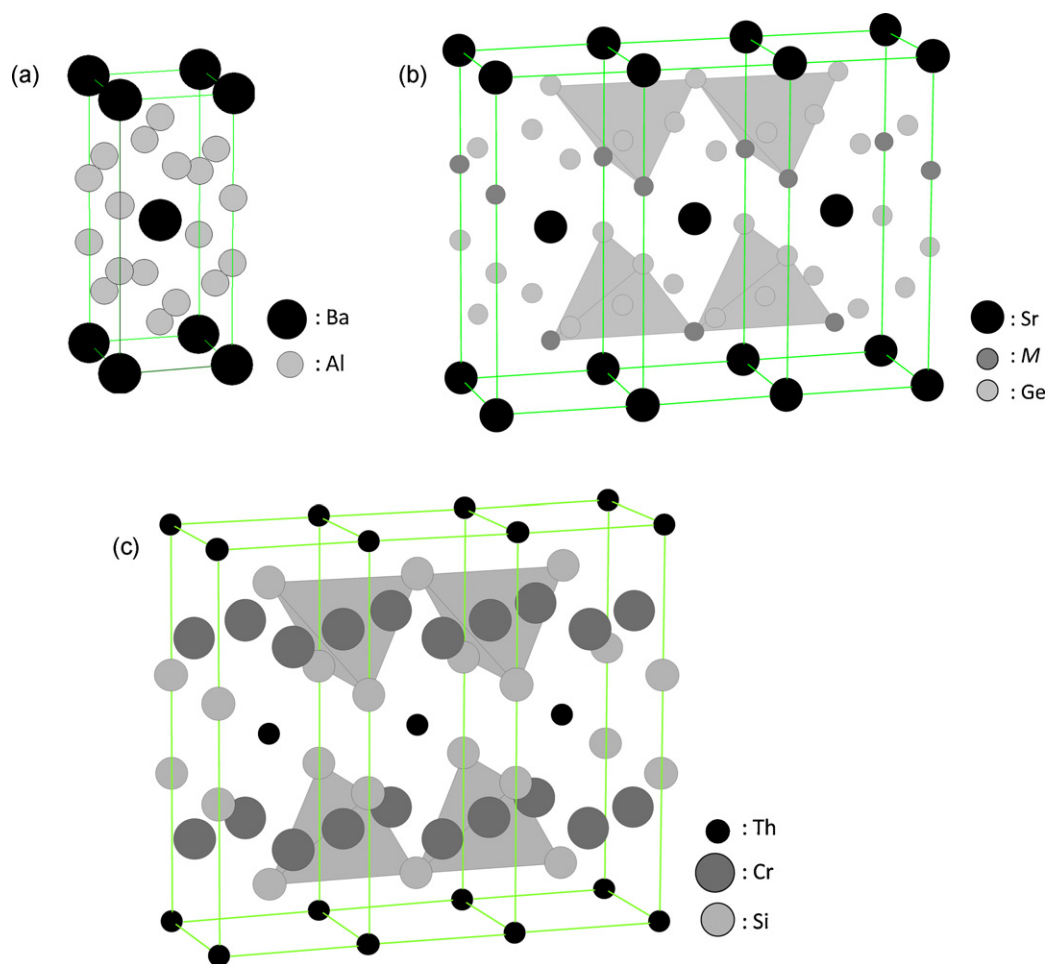


Fig. 2. Crystal structure of (a) BaAl₄, together with its derivative (b) BaNiSn₃-type SrMGe₃ ($M = \text{Ir, Pd and Pt}$) with Ge(2)M₂Ge(1)₂ tetrahedra and (c) ThCr₂Si₂ with CrSi₄ tetrahedra.

and SrNiGe₃, we could not pick up single crystalline samples. The lattice parameters of those two intermetallics suggest that they are likely to take the same BaNiSn₃-type structure and Rietveld structural refinements using the XRD patterns of those polycrystalline samples are under way.

Table 3 lists the lattice parameters, atomic radii r_i [17], interatomic distances d_{i-j} and their corresponding relative dilatations Δ_{i-j} in SrMGe₃ ($M = \text{Ir, Pd and Pt}$), together with ThCr₂Si₂-type SrM₂Ge₂ ($M' = \text{Ir, Ni and Pd}$) for comparison [10,11]. Here Δ_{i-j} is defined as $\Delta_{i-j} = (d_{i-j} - \sum r_i) / \sum r_i$. The axial ratio c/a is almost the same for BaNiSn₃-type ternary germanides AMGe₃, ranging from 2.24 in BaPtGe₃ [16] to 2.30 in SrPdGe₃. Small atoms such as Y at the 2a site in AMGe₃ have not been reported so far and the decrease in size of the atom A at the 2a site in the AMGe₃ turns the structure to cubic Yb₃Rh₄Sn₁₃-type with similar composition [15].

On the other hand, in the ThCr₂Si₂-type AM₂Ge₂ the c/a ratio ranges from 2.29 in LaPd₂Ge₂ to 2.73 in YMn₂Ge₂ [18,19]. The decrease in size of the atom A at the 2a site in AM₂Ge₂ does not always decrease the c -axis parameter. This relatively large range of the c/a ratio in ThCr₂Si₂-type, compared with BaNiSn₃-type, is associated with the ability of accommodating various combinations of three atoms. Hence, a huge number of intermetallic compounds with this structure are observed. On the other hand, the small range of the c/a ratio in BaNiSn₃-type means a smaller number of intermetallics with this structure.

These ThCr₂Si₂- and BaNiSn₃-type structures have layers of M'Ge₄ and Ge(2)M₂Ge(1)₂ tetrahedra, respectively, as shown in Fig. 2. These layers are connected with Ge–Ge and M–Ge(1) bonds.

The $\Delta_{M-\text{Ge}(1)}$ in SrMGe₃ is almost equal and small, suggesting strong interactions between the layers of the Ge(2)M₂Ge(1)₂ tetrahedra. On the other hand, the $\Delta_{\text{Ge}-\text{Ge}}$ in SrM₂Ge₂ is scattered and large, suggesting weak interactions between the layers. Other ThCr₂Si₂-type ternary germanides also show such scattered $\Delta_{\text{Ge}-\text{Ge}}$, ranging from –11.3 in YCu₂Ge₂ to +4.2 in SrRu₂Ge₂ [19,20]. Therefore, the difference of the interactions between the layers causes the difference of range in the c/a ratio and hence the number of intermetallics for these two structures.

Fig. 3 shows temperature-dependent DC magnetization curves measured in both zero-field cooling (ZFC) and field cooling (FC) modes for polycrystalline samples of SrMGe₃ ($M = \text{Ir, Pd and Pt}$). The SrPdGe₃ sample does not show diamagnetic signals down to 1.8 K, indicating that the sample is not superconducting. This is in contrast to BaAl₄-derivative ThCr₂Si₂-type SrPd₂Ge₂ with a T_c of 3.0 K [10].

On the other hand, the other two samples SrIrGe₃ and SrPtGe₃ show a diamagnetic signal at around 5 K with a broad transition. The magnitude of magnetic shielding signal after being corrected for demagnetization effects is below 1% of that estimated for perfect diamagnetism. Taking into account the volume fraction of SrIrGe₃ and SrPtGe₃ phases in the samples from XRD analyses shown in Fig. 1, we conclude that neither of SrIrGe₃ nor SrPtGe₃ is superconducting down to 1.8 K. In a binary system of Ir–Ge, IrGe shows superconductivity with a T_c of 4.7 K [21]. In ternary systems of A–Ir–Ge, some superconducting materials such as La₃Ir₂Ge₂ with a T_c of 4.7 K [22] have been reported so far. On the other hand, in a Pt–Ge system, PtGe shows superconductivity with a T_c of 0.4 K [23].

Table 3

Lattice parameters, atomic radii, interatomic distances and their corresponding relative dilatations in BaNiSn₃-type SrMGe₃ (*M* = Ir, Pd and Pt) and ThCr₂Si₂-type SrM'₂Ge₂ (*M'* = Ir, Ni and Pd) [10,11]. Here Δ_{i-j} is relative dilatations of the interatomic distances from atomic radii, defined as $\Delta_{i-j} = (d_{i-j} - \sum r_i) / \sum r_i$ where d_{i-j} and r_i denote interatomic distances of *i*–*j* and atomic radius of *i*, respectively. The radius of Ge used is 0.137 nm [17].

Compound	SrIrGe ₃	SrPdGe ₃	SrPtGe ₃	SrIr ₂ Ge ₂	SrNi ₂ Ge ₂	SrPd ₂ Ge ₂
Lattice parameters						
<i>a</i> (nm)	0.44653(1)	0.44623(2)	0.44778(2)	0.41916(2)	0.41801(1)	0.44088(2)
<i>c</i> (nm)	1.00909(5)	1.02737(7)	1.01366(9)	1.0674(1)	1.02409(6)	1.01270(8)
<i>c/a</i>	2.2598	2.3023	2.2637	2.5465	2.4499	2.2970
Cell volume (nm ³)	0.20120(1)	0.20457(2)	0.20325(2)	0.18753(2)	0.17894(1)	0.19684(2)
Atomic radius of <i>M</i> (<i>M'</i>) (nm)	0.135	0.137	0.139	0.135	0.125	0.137
<i>d</i> _{Sr–M} (nm)	0.3499(1)	0.34758(8)	0.34901(8)			
	0.3539(3)	0.3679(2)	0.3600(2)			
<i>d</i> _{Sr–Ge(1)} (nm)	0.3341(2)	0.3378(2)	0.3344(2)			
	0.3397(2)	0.3427(2)	0.3420(1)			
<i>d</i> _{Sr–Ge(2)} (nm)	0.3279(1)	0.33202(8)	0.33209(8)			
<i>d</i> _{M–Ge(1)} (nm)	0.24688(9)	0.25070(7)	0.25021(6)			
$\Delta_{M-Ge(1)}$ (%)	–9.2	–8.5	–9.3			
<i>d</i> _{M–Ge(2)} (nm)	0.2391(3)	0.2491(2)	0.2470(2)			
<i>d</i> _{Ge(1)–Ge(2)} (nm)				0.2766(3)	0.28332(7)	0.2626(1)
Δ_{Ge-Ge} (%)				+0.9	+3.4	–4.2

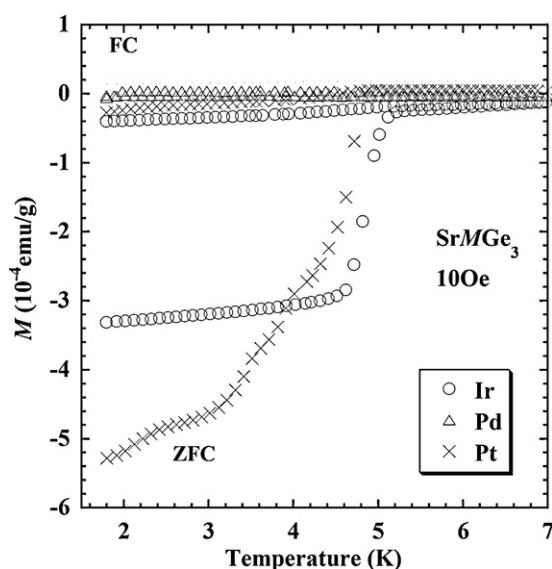


Fig. 3. Temperature-dependent DC magnetization curves for polycrystalline samples of SrMGe₃ (*M* = Ir, Pd and Pt). The data were recorded in both zero-field cooling (ZFC) and field cooling (FC) modes. The applied field was 10 Oe.

In the corresponding ternary system, SrPt₄Ge₁₂ shows superconductivity with a *T*_c of 5.4 K [24]. Therefore, the superconductivity observed in our samples is possibly due to the presence of such binary or ternary germanides.

SrIrGe₃ does not show superconductivity down to 1.8 K, as mentioned above. However, the color of the SrIrGe₃ sample is slightly red, the same as superconducting SrPd₂Ge₂. Therefore, SrIrGe₃ may show superconductivity below 1.8 K.

4. Conclusions

We have prepared ternary germanides SrMGe₃ (*M* = Rh, Ir, Ni, Pd and Pt) by arc melting. The germanides have a body-centered tetragonal unit cell with BaNiSn₃-type structure with the space

group *I4mm* (No. 107) for *M* = Ir, Pd and Pt, determined by single-crystal X-ray diffraction. The other two are also likely to take the same structure. DC magnetization measurements indicated that neither of them shows superconductivity down to 1.8 K.

References

- [1] Z. Ban, M. Sikirica, Acta Crystallogr. 18 (1965) 594–599.
- [2] G.W. Hull, J.H. Wernick, T.H. Geballe, J.V. Waszczak, J.E. Bernardini, Phys. Rev. B 24 (1981) 6715–6718.
- [3] M. Francois, G. Venturini, J.F. Maréché, B. Malaman, B. Roques, J. Less-Common Met. 113 (1985) 231–237.
- [4] R.J. Cava, H. Takagi, B. Batlogg, H.W. Zandbergen, J.J. Krajewski, W.F. Peck Jr., R.B. van Dover, R.J. Felder, T. Siegrist, K. Mizuhashi, J.O. Lee, H. Eisaki, S.A. Carter, S. Uchida, Nature 367 (1994) 146–148.
- [5] R.J. Cava, H. Takagi, H.W. Zandbergen, J.J. Krajewski, W.F. Peck Jr., T. Siegrist, B. Batlogg, R.B. van Dover, R.J. Felder, K. Mizuhashi, J.O. Lee, H. Eisaki, S. Uchida, Nature 367 (1994) 252–253.
- [6] R.J. Cava, B. Batlogg, T. Siegrist, J.J. Krajewski, W.F. Peck Jr., S. Carter, R.J. Felder, H. Takagi, R.B. van Dover, Phys. Rev. B 49 (1994) 12384–12387.
- [7] H. Fujii, S. Ikeda, T. Kimura, S.-i. Arisawa, K. Hirata, H. Kumakura, K. Kadowaki, K. Togano, Jpn. J. Appl. Phys. 33 (1994) L590–L593.
- [8] M. Rotter, M. Tegel, D. Johrendt, Phys. Rev. Lett. 101 (2008) 107006(1)–107006(4).
- [9] K. Sasmal, B. Lv, B. Lorenz, A.M. Guloy, F. Chen, Y.-Y. Xue, C.-W. Chu, Phys. Rev. Lett. 101 (2008) 107007(1)–107007(4).
- [10] H. Fujii, A. Sato, Phys. Rev. B 79 (2009) 224522(1)–224522(5).
- [11] H. Fujii, A. Sato, J. Alloys Compd. 487 (2009) 198–201.
- [12] G.M. Sheldrick, SADABS. Program for Empirical Absorption Correction of Area Detector Data, Univ. of Göttingen, 1996.
- [13] Bruker, SMART, SAINT and SADABS, Bruker AXS Inc., Madison, WI, USA, 2001.
- [14] G.M. Sheldrick, Acta Crystallogr. A64 (2008) 112–122.
- [15] G. Venturini, M. Méot Meyer, B. Malaman, B. Roques, J. Less-common Met. 113 (1985) 197–204.
- [16] R. Demchyna, Yu. Prots, W. Schnelle, U. Burkhardt, U. Schwarz, Z. Kristallogr. NCS 221 (2006) 109–111.
- [17] E. Teatum, K. Gschneider, J. Waber, in: W.B. Pearson (Ed.), The Crystal Chemistry and Physics of Metals and Alloys, Wiley, New York, 1972, p. 151.
- [18] G. Venturini, B. Malaman, J. Alloys Compd. 235 (1996) 201–209.
- [19] G. Venturini, B. Malaman, B. Roques, J. Solid State Chem. 79 (1989) 136–145.
- [20] J. González, R. Kessens, H.U. Schuster, Z. Anorg. Allg. Chem. 619 (1993) 13–16.
- [21] B.T. Matthias, Phys. Rev. 92 (1953) 874–876.
- [22] D.X. Li, S. Nimori, Y. Homma, Y. Shiokawa, A. Tobo, H. Onodera, Y. Haga, Y. Onuki, J. Appl. Phys. 97 (2005) 073903(1)–073903(6).
- [23] B.T. Matthias, T.H. Geballe, V.B. Compton, Rev. Mod. Phys. 35 (1–22) (1963) 414.
- [24] R. Gumeniuk, W. Schnelle, H. Rosner, M. Nicklas, A. Leithe-Jasper, Yu. Grin, Phys. Rev. Lett. 100 (2008) 017002(1)–017002(4).